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Key indicators

Single-crystal X-ray study T = 293 KMean $\sigma(\text{C}-\text{C}) = 0.008 \text{ Å}$ R factor = 0.065 wR factor = 0.190 Data-to-parameter ratio = 14.2

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e. GBR12909, $C_{28}H_{34}F_2N_2O$, is one of the earliest agents found to have high affinity and selectivity for the dopamine transporter (DAT). The triclinic crystal, of the hydrochloride salt of GBR12909, $C_{28}H_{34}F_2N_2O^{2+}\cdot 2Cl^-$, exhibited nonmerohedral twinning.

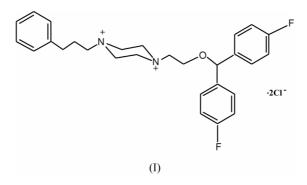
Twinned 1-{2-[bis(4-fluorophenyl)methoxy]ethyl}-4-(3-

phenyl)propyl)piperazinium chloride (GBR12909)

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Comment

Cocaine is one of the most widely abused drugs in the world. The reinforcing and locomotor stimulating properties of cocaine are said to be mainly mediated by the binding of cocaine to the dopamine transporter (DAT) and subsequent blockade of dopamine re-uptake into presynaptic terminals, resulting in increased neurotransmission in the mesolimbic dopaminergic system (Kuhar et al., 1991; Rothman et al., 1989). Studies have been aimed at the discovery and development of potent and selective DAT ligands, (Zhang et al., 2000; Newman et al., 1994) that may reduce the effects of cocaine and, therefore, could be used for the treatment of cocaine abuse (Glowa et al., 1995a,b). GBR12909, (I), a potent if not overly selective DAT ligand, is now being examined in man for its toxicity and efficacy prior to exploration of its potential as a cocaine-treatment agent. The X-ray structure of GBR12909 will provide some of the necessary information for structure-activity studies that might lead to a new generation of more potent and selective cocaine-treatment agents.



GBR12909 hydrochloride crystallizes with two molecules per asymmetric unit. The two independent molecules are almost identical, the only difference being the relative twist of the two fluorophenyl groups (Fig. 1). The difference between the C1B-C10-C1A-C2A torsion angle and the corresponding torsion in the second molecule in the asymmetric unit is approximately 16°. The central six-membered ring has a normal chair conformation. Both N atoms in the ring are positively charged and hydrogen bond to a Cl ion (see Table 1). There are no other intermolecular interactions. Both aromatic moieties are extended away from the central ring and from

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organic papers

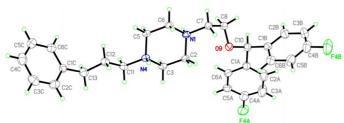


Figure 1

View of GBR12909 with 20% probability ellipsoids, showing one of the two independent molecules in the asymmetric unit.

each other. The 4-fluorophenyl rings are approximately perpendicular to one another (angle between the ring planes of 88.5 (1) and 88.6 (2) $^{\circ}$ for the two independent molecules). The third aromatic ring is approximately perpendicular to the plane formed by N4, C11, C12 and C13 [82.3 (3) and 82.2 (2)° for the two independent molecules].

Experimental

The sample of GBR12909 hydrochloride was obtained from the National Institutes of Health. Crystals were grown at NRL by slow evaporation from a mixture of 2-propanol and ethyl acetate.

 $h = -10 \rightarrow 10$

 $k = -16 \rightarrow 14$

 $l = -26 \rightarrow 25$

56 standard reflections

intensity decay: none

 $+ 2F_c^2)/3$

Crystal data

$C_{28}H_{34}F_2N_2O^{2+}\cdot 2Cl^{-}$	Z = 4
$M_r = 523.47$	$D_x = 1.242 \text{ Mg m}^{-3}$
Triclinic, $P\overline{1}$	Cu $K\alpha$ radiation
a = 8.5449 (4) Å	Cell parameters from 5787
b = 14.3779 (7) Å	reflections
c = 23.2700 (12) Å	$\theta = 3.1-61.4^{\circ}$
$\alpha = 83.131 \ (3)^{\circ}$	$\mu = 2.38 \text{ mm}^{-1}$
$\beta = 80.554 \ (3)^{\circ}$	T = 293 (2) K
$\gamma = 89.963 \ (3)^{\circ}$	Plate, colorless
V = 2799.4 (2) Å ³	$0.40 \times 0.35 \times 0.02 \text{ mm}$
Data collection	
Bruker SMART 6000	6598 reflections with $I > 2\sigma(I)$
diffractometer	$\theta_{\rm max} = 67.2^{\circ}$
	1 10 10

diffractometer ω scans Absorption correction: multi-scan (SADABS; Bruker, 2001) $T_{\min} = 0.386, T_{\max} = 0.953$ 9044 measured reflections 9044 independent reflections

Refinement

Refinement on F^2 $w = 1/[\sigma^2(F_o^2) + (0.128P)^2]$ $R[F^2 > 2\sigma(F^2)] = 0.065$ where $P = (F_o)$ $wR(F^2) = 0.190$ $(\Delta/\sigma)_{\rm max} = 0.047$ $\Delta \rho_{\rm max} = 0.42 \text{ e } \text{\AA}^{-3}$ S = 0.97 $\Delta \rho_{\rm min} = -0.26 \text{ e } \text{\AA}^{-3}$ 9044 reflections Extinction correction: SHELXL97 635 parameters H atoms treated by a mixture of Extinction coefficient: 0.0015 (3) independent and constrained refinement

Table 1

Hydrogen-bonding geometry (Å, °).

$D - H \cdots A$	$D-\mathrm{H}$	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$N1-H1A\cdots Cl4$	0.91	2.14	3.034 (3)	166
$N4-H4A\cdots Cl3$	0.91	2.11	3.012 (3)	174
$N1' - H1'A \cdots Cl2$	0.91	2.13	3.027 (3)	167
$N4' - H4'A \cdots Cl1$	0.91	2.11	3.018 (3)	173

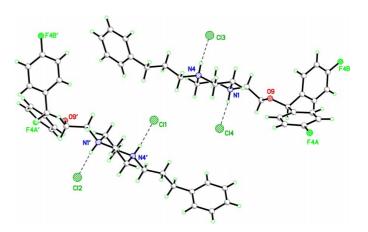


Figure 2

The asymmetric unit, showing each independent molecule hydrogen bonded to two Cl atoms.

The GBR12909 hydrochloride crystals exhibited non-merohedral twinning. Program ROTAX, found in the CRYSTALS (Watkin, 2001) 0,0.4,-1). Program ROTWIN (Young, 2001) was used to generate a new reflection file for refinement with SHELXTL (Bruker, 2001). Without twinning the refinement converged to an R factor of 0.094. Adding the single twin reduced the R factor to 0.065.

Data collection: SMART (Bruker, 2001); cell refinement: SAINT (Bruker, 2001); data reduction: SAINT; program(s) used to solve structure: SHELXTL (Sheldrick, 1997); program(s) used to refine structure: SHELXTL (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 2001); software used to prepare material for publication: SHELXTL (Bruker, 2001).

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